

# A SPIRO-BIS- $\gamma$ -LACTONE GLUCOSIDE FROM *VIBURNUM DILATATUM*

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**Key Word Index**—*Viburnum dilatatum*, Caprifoliaceae, spiro-bis- $\gamma$ -lactone glucoside dilaspirolactone

**Abstract**—A new spiro-bis- $\gamma$ -lactone glucoside was isolated from the leaves of *Viburnum dilatatum* and its structure was elucidated on the basis of spectral and chemical evidence

## INTRODUCTION

The deciduous shrub *Viburnum dilatatum* is widely distributed in Japan. The fruits are used as a spice in wine and pickles. Previous work on the plant has yielded a number of known compounds from the flowers and fruits [1, 2]. We have now examined the methanolic extract of the leaves. As a result a new glucoside (1), which is named dilaspirolactone, as well as sitosterol, ursolic acid, sitosteryl  $\beta$ -D-glucoside, isoquercitrin and *p*-hydroxyphenyl  $\beta$ -D-alloside [3] have been isolated.

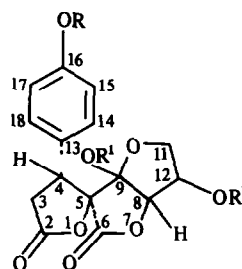
## RESULTS AND DISCUSSION

Compound 1 was crystallized as prisms, mp 165–167° with a molecular formula  $C_{21}H_{24}O_{13} \cdot H_2O$ . It exhibited typical IR absorption bands of  $\gamma$ -lactones at 1810 and 1780  $cm^{-1}$ , the former band indicating an electronegative substituent in the  $\gamma$ -position. Further absorption bands at 3300  $cm^{-1}$ , and 1610, 1600, 1515 and 840  $cm^{-1}$  showed a hydroxyl group and a *p*-substituted phenyl group, respectively. The  $^1H$  NMR spectrum indicated an ABX system at  $\delta$  3.35, 3.51 and 5.47 ( $J = 9, 12$  and 17 Hz) arising from a  $-CH_2CH-$  group in addition to signals of the *p*-substituted phenyl group at  $\delta$  7.02 and 7.52 ( $A_2B_2$ ,  $J = 10$  Hz). Acetylation of 1 with acetic anhydride in pyridine gave a hexa-acetate (2),  $C_{33}H_{36}O_{19}$ , and a penta-acetate (3),  $C_{31}H_{34}O_{18}$ . The IR spectrum of 3 showed an absorption band typical of a tertiary hydroxyl group at 3480  $cm^{-1}$ .

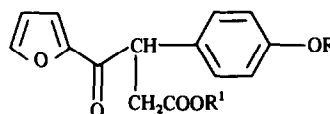
On hydrolysis with 2 N HCl, compound 1 afforded a carboxylic acid (4), mp 133° with a molecular formula  $C_{14}H_{12}O_5$  and D-glucose. The IR spectrum of 4 showed absorption bands for hydroxyl (3400  $cm^{-1}$ ), carboxylic (2750–2400 and 1710  $cm^{-1}$ ) and conjugated carbonyl (1670  $cm^{-1}$ ) functionalities together with a *p*-substituted phenyl group (1620, 1600, 1515 and 820  $cm^{-1}$ ). An absorption band at 1570  $cm^{-1}$  suggested that a furan ring was newly produced. The  $^1H$  NMR spectrum also indicated the presence of the furan group. Signals at  $\delta$  6.52 (1H, *dd*,  $J = 2$  and 3 Hz), 7.22 (1H, *d*,  $J = 3$  Hz) and 7.67 (1H, *d*-like,  $J = 2$  Hz) were in good agreement with those observed in the  $\alpha$ -furoyl group of furoin [4]. An ABX system at  $\delta$  2.59, 3.24 and 4.78 ( $J = 5, 10$  and 17 Hz) was due to a  $-CH_2CH-$  group. The presence of the *p*-substituted phenyl group was confirmed by signals at  $\delta$  6.74 and 7.15 (2H each,  $A_2B_2$ ,  $J = 8$  Hz). The above results suggested that compound 4 was 3-( $\alpha$ -furoyl)-3-(*p*-

hydroxyphenyl)-propionic acid, which was also supported by the  $^{13}C$  NMR spectrum of 4.

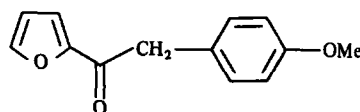
Acid methanolysis of 1 yielded a methyl ester (5),  $C_{15}H_{14}O_5$ , and a methyl ether (6), mp 261–262°,  $C_{16}H_{16}O_8$ . The  $^1H$  NMR spectrum of 5 was very similar to that of 4 except for an additional three-proton singlet at  $\delta$  3.72 due to a carbomethoxyl group. Therefore, compound 5 could be deduced to be methyl 3-( $\alpha$ -furoyl)-3-(*p*-



- 1 R = R<sup>1</sup> = H, R<sup>2</sup> = Glu
- 2 R = R<sup>1</sup> = Ac, R<sup>2</sup> = GluAc<sub>4</sub>
- 3 R = Ac, R<sup>1</sup> = H, R<sup>2</sup> = GluAc<sub>4</sub>
- 4 R = R<sup>2</sup> = H, R<sup>1</sup> = Me



- 4 R = R<sup>1</sup> = H
- 5 R = H, R<sup>1</sup> = Me
- 7 R = R<sup>1</sup> = Me
- 9 R = Me, R<sup>1</sup> = Et



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hydroxyphenyl)-propionate, which was confirmed as the corresponding methyl ether (7) by preparing it from furan Acylation of furan with *p*-methoxyphenylacetic acid in the presence of phosphorus pentoxide gave 2-(*p*-methoxyphenyl)-acetylfuran (8). Condensation of 8 with ethyl bromoacetate in benzene containing sodium resulted in formation of an ethyl ester (9). The ethyl ester 9 was subjected to hydrolysis with 1N NaOH followed by treatment with diazomethane to give a methyl ester, whose IR and <sup>1</sup>H NMR spectra were identical with those of 7 obtained by methylation of 5 with diazomethane. On the other hand, an absorption band at 1800 cm<sup>-1</sup> in the IR spectrum of 6 showed that the  $\gamma$ -lactone in 1 remained intact. The IR spectrum also indicated absorption bands 3590 (OH), 3350 (phenolic OH), 1610, 1590, 840 and 815 cm<sup>-1</sup> (*p*-substituted phenyl group). The structure of the aglycone 6 could be established from the <sup>1</sup>H NMR data. A three-proton singlet assignable to a methoxyl group appeared at  $\delta$  3.71.

Thus, one oxygen atom remained to be accounted for, and the molecular formula suggested that compound 6 had a furanose moiety (C-ring) as a third ring. This was reminiscent of piptoside having two spiro-joined  $\gamma$ -lactone rings, one of which was fused to a ketofuranose ring [5]. The methoxyl group was therefore formed by methylation of a hemiacetal hydroxyl group at C-9. An ABX system at  $\delta$  4.27, 4.47 and 4.65 ( $J = 4, 6.4$  and 10 Hz) was assigned to the three protons on C-ring (H-11  $\times$  2 and H-12). Thus, the remaining one aliphatic hydroxyl group to be assigned could be placed at C-12, and its  $\beta$ -configuration was assumed from the coupling constant ( $J = 0$  Hz) between the C-12 and C-13 protons. Another ABX system at  $\delta$  3.15, 3.59 and 4.46 ( $J = 8.6, 13$  and 17 Hz) was very close to that of the A-ring in leucodrin [6], suggesting that the *p*-hydroxyphenyl group was located at C-4 and had a  $\alpha$ -configuration.

The glucose was attached to the hydroxyl group at C-12 and the  $W_{1/2}$  value (10 Hz) of the anomeric proton at  $\delta$  5.71 in the <sup>1</sup>H NMR spectrum of 1 suggested that the glycosidic linkage must be  $\beta$  in 1.

To establish the absolute structure of 1, it was methylated with diazomethane followed by oxidation with sodium periodate yielding fairly racemized (-)-*p*-methoxyphenylsuccinic acid, mp 198–200°,  $[\alpha]_D -15^\circ$  [6]. Therefore, dilaspirolactone was shown to have the absolute structure 1.

A series of natural products containing the 1,7-dioxo-2,6-dioxospiro [4,4] nona skeleton has so far been isolated, leucodrin, leudrin and leucoglycodrin from *Leucadendron* sp (Proteaceae) [6, 7], conocarpin and two closely related ring-A-opened lactones, conocarpic acid and reflexin from *Luecospermum* sp (Proteaceae) [8, 9], and piptoside from *Piptocalys moorei* (Trimeniaceae) [5]. This is the first example of the spiro-bis- $\gamma$ -lactone isolated from the Caprifoliaceae.

#### EXPERIMENTAL

**Extraction and isolation.** Plant material was collected in Kagoshima city and identified by Dr S Sako (Herbarium sample No. 6). Fresh leaves of *V. dilatatum* (2.9 kg) were extracted with MeOH (13 l  $\times$  2). After concentration of the combined MeOH solns, H<sub>2</sub>O was added and the insoluble material filtered

off. The filtrate was extracted with Et<sub>2</sub>O and then EtOAc. The Et<sub>2</sub>O extract was evaporated to give a dark green residue (25 g), which was chromatographed on a column of silica gel, eluting with CHCl<sub>3</sub>-MeOH with increasing proportions of MeOH. From the fractions eluted with CHCl<sub>3</sub>, sitosterol (326 mg) and ursolic acid (87 mg) were obtained. The fractions eluted with CHCl<sub>3</sub>-MeOH (9/1) gave sitosteryl  $\beta$ -D-glucoside (98 mg). The EtOAc extract was evaporated to give a reddish brown residue (31 g), which was subjected to CC on silica gel with CHCl<sub>3</sub>-MeOH with increasing proportions of MeOH. Elution with CHCl<sub>3</sub>-MeOH (17/3) gave isoquercitrin (70 mg) and *p*-hydroxyphenyl  $\beta$ -D-alloside (110 mg) successively. Dilaspirolactone 1 (941 mg) was obtained from the fractions eluted with CHCl<sub>3</sub>-MeOH (5/1).

**Dilaspirolactone 1.** Prisms from Me<sub>2</sub>CO-MeOH, mp 165–167°,  $[\alpha]_D -17.6^\circ$  (MeOH,  $c$  0.5), UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 202 (7000), 227 (6800), 276 (1100), 284 (8700), IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup> 3300, 1810, 1780, 1610, 1600, 1515, 900, 840, <sup>1</sup>H NMR (200 MHz, C<sub>5</sub>D<sub>5</sub>N)  $\delta$  3.35, 3.51 and 5.47 (ABX,  $J_{A,B} = 17$ ,  $J_{A,X} = 9$  and  $J_{B,X} = 12$  Hz, H-3  $\times$  2, and H-4), 4.20–4.80 (10H, m), 5.72 (1H,  $W_{1/2} = 10$  Hz, sugar H-1), 7.02 and 7.32 (A<sub>2</sub>B<sub>2</sub>,  $J = 8$  Hz), <sup>13</sup>C NMR (22.6 MHz, C<sub>5</sub>D<sub>5</sub>N)  $\delta$  34.0 (C-3), 44.3 (C-4), 61.8 (C'-6), 71.0 (C'-4), 74.2\* (C'-2), 74.7\* (C'-3), 77.1 (C-11), 78.3† (C'-5), 78.6† (C-8), 88.6 (C-12), 90.5 (C-5), 97.5 (C'-1), 108.7 (C-9), 116 (C-15 and C-17), 123.6 (C-13), 130.8 (C-14 and C-18), 159.2 (C-16), 171.8 (C-6), 174.8 (C-2), MS  $m/z$  (rel int) no [M]<sup>+</sup>, 322 (3), 304 (8), 147 (63), 120 (91), 73 (50), 61 (100) (Found C, 51.23, H, 5.61%). Calc for C<sub>21</sub>H<sub>24</sub>O<sub>13</sub> H<sub>2</sub>O C, 51.20, H, 5.22%). Compound 1 (40 mg) was acetylated with Ac<sub>2</sub>O in pyridine at 5° CC of the crude product gave 2 (24 mg) and 3 (10 mg). Compound 2, an amorphous powder, IR  $\nu_{\text{max}}^{\text{film}}$  cm<sup>-1</sup> 1820, 1760, 1615, 1515, 1220, <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  2.00 and 2.04 (3H  $\times$  4, s, OAc), 2.32 (3H, s, OAc), 2.88–3.52 (2H, m), 3.84–4.20 (2H, m), 4.20–4.90 (2H, m), 5.10–5.48 (5H, m), 7.24 and 7.40 (A<sub>2</sub>B<sub>2</sub>,  $J = 8$  Hz), MS  $m/z$  736 [M]<sup>+</sup> (Found C, 54.27, H, 5.32%). Calc for C<sub>33</sub>H<sub>36</sub>O<sub>14</sub> C, 53.80, H, 4.93%). Compound 3, an amorphous powder, IR  $\nu_{\text{max}}^{\text{film}}$  cm<sup>-1</sup> 3480, 1810, 1750, 1605, 1510, 1225, <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  2.06, 2.07, 2.12, 2.14, 2.30 (3H each, s), 2.94 and 3.51 (AB of ABX,  $J_{A,B} = 17$  Hz,  $J_{A,X} = 9$  Hz,  $J_{B,X} = 13$  Hz, H-3  $\times$  2), 3.52–3.84 (2H, m), 3.88–3.94 (1H, m), 4.01 (1H, s, OH), 4.20 (1H, dd,  $J = 3$  and 13 Hz, H'-6), 4.30 (1H, dd,  $J = 2$  and 13 Hz, H'-6), 4.56–4.66 (2H, m), 4.92–5.04 (1H, m), 5.04–5.28 (3H, m), 7.14 and 7.36 (A<sub>2</sub>B<sub>2</sub>,  $J = 9$  Hz), MS  $m/z$  694 [M]<sup>+</sup>.

**Hydrolysis of 1.** A soln of 1 (21 mg) in 2 N HCl (1 ml) was refluxed for 2 hr. The reaction soln was extracted with Et<sub>2</sub>O. The Et<sub>2</sub>O extract was evaporated and recrystallized from CHCl<sub>3</sub>-MeOH to give pale yellow needles of 4 (7 mg), mp 133°, UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 220 (15000), 270 (21000), IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup> 3450, 2750–2400, 1710, 1670, 1620, 1600, 1570, 1515, 980, 840, 820. <sup>1</sup>H NMR [90 MHz, (CD<sub>3</sub>)<sub>2</sub>CO]  $\delta$  2.59, 3.24 and 4.78 (ABX,  $J_{A,B} = 17$  Hz,  $J_{A,X} = 5$  Hz and  $J_{B,X} = 10$  Hz, H-2  $\times$  2 and H-3), 6.52 (1H, dd,  $J = 2$  and 3 Hz, furan H-4), 6.74 and 7.15 (A<sub>2</sub>B<sub>2</sub>,  $J = 8$  Hz, Ar-H), 7.22 (1H, d,  $J = 3$  Hz, furan H-3), 7.67 (1H, br d,  $J = 2$  Hz, furan H-5), <sup>13</sup>C NMR [22.6 MHz, (CD<sub>3</sub>)<sub>2</sub>CO]  $\delta$  37.7 (CH<sub>2</sub>), 49.2 (CH), 112.9 (furan C-4), 116.4 (Ar C-3 and C-5), 118.5 (furan C-3), 130.1 (Ar C-1, C-2 and C-6), 147.5 (furan C-5), 153.1 (Ar C-4), 157.5 (furan C-2), 172.9 (COOH), 187.9 (CO), MS  $m/z$  (rel int) 260 [M]<sup>+</sup> (43), 165 (67), 123 (100), 120 (37), 95 (44) (Found  $m/z$  260.0776. Calc for C<sub>14</sub>H<sub>12</sub>O<sub>5</sub>  $m/z$  260.0685).

**Methanolysis of 1.** To a soln of 1 (130 mg) in dry MeOH (3 ml), was added three drops of conc HCl and the mixture refluxed for 6 days. The reaction mixture was diluted with H<sub>2</sub>O and extracted with Et<sub>2</sub>O. The extract was washed with H<sub>2</sub>O and dried (Na<sub>2</sub>SO<sub>4</sub>). The crude product was recrystallized from CHCl<sub>3</sub>-MeOH to afford 6 (12 mg), prisms, mp 261–262°,  $[\alpha]_D -33.3^\circ$  (MeOH,  $c$  0.33), IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup> 3590, 3350, 1800, 1610,

\*, † Assignments may be interchanged

1590, 895, 840, 815,  $^1\text{H NMR}$  (400 MHz,  $\text{C}_5\text{D}_5\text{N}$ )  $\delta$  3.15, 3.59 and 4.49 (ABX,  $J_{\text{AB}} = 17$  Hz,  $J_{\text{AX}} = 8.6$  Hz, and  $J_{\text{BX}} = 13$  Hz, H-2  $\times$  2 and H-3), 3.62 (1H, s, OH), 3.71 (3H, s, OMe), 4.27, 4.47 and 4.67 (ABX,  $J_{\text{AB}} = 10$  Hz,  $J_{\text{AX}} = 6.4$  Hz and  $J_{\text{BX}} = 4$  Hz, H-11  $\times$  2 and H-10), 7.15 and 7.53 ( $\text{A}_2\text{B}_2$ ,  $J = 8.5$  Hz, Ar-H), MS  $m/z$  (rel int) 336 [ $\text{M}$ ] $^+$  (15), 318 (13), 203 (13), 147 (15), 120 (100), 99 (79), 91 (31) (Found  $m/z$  336.0846 Calc for  $\text{C}_{16}\text{H}_{16}\text{O}_8$ ,  $m/z$  336.0845) The mother soln was evapd and subjected to CC on silica gel Elution with  $\text{CHCl}_3$ -MeOH (99:1) gave a solid 5 (11 mg),  $[\alpha]_{\text{D}} -4^\circ$  (MeOH,  $c$  0.75), UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ), 222 (8700), 272 (12900), IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  3400, 1730, 1665, 1610, 1590, 1560, 1510, 960, 900, 880, 835, 810,  $^1\text{H NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\delta$  2.77, 3.40 and 4.94 (ABX,  $J_{\text{AB}} = 20$  Hz,  $J_{\text{AX}} = 6$  Hz and  $J_{\text{BX}} = 10$  Hz, H-2  $\times$  2 and H-3), 3.72 (3H, s, COOMe), 6.63 (1H,  $dd$ ,  $J = 2$  and 4 Hz, furan H-4), 6.96 and 7.39 ( $\text{A}_2\text{B}_2$ ,  $J = 9$  Hz), 7.40 (1H,  $d$ ,  $J = 4$  Hz, furan H-3), 7.76 (1H,  $m$ , furan H-3), MS  $m/z$  274 [ $\text{M}$ ] $^+$  (Found  $m/z$  274.0843 Calc for  $\text{C}_{15}\text{H}_{14}\text{O}_5$ ,  $m/z$  274.0841)

2-(*p*-Methoxyphenyl)-acetyl-furan (8)  $\text{P}_2\text{O}_5$  (4.2 g) was weighed into benzene (16 ml) containing furan (2 g) and the soln was slowly added to *p*-methoxyphenylacetic acid (4.88 g) The reaction mixture was refluxed for 4 hr and the benzene layer washed with dil NaOH and  $\text{H}_2\text{O}$  CC of the crude product on silica gel with  $\text{CHCl}_3$ -hexane (1:1) gave the acetate 8 (200 mg), an oil, IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  3125, 2850, 1670, 1610, 1565, 1510, 880, 820,  $^1\text{H NMR}$  (60 MHz,  $\text{CDCl}_3$ )  $\delta$  3.77 (3H, s), 4.05 (2H, s), 6.50 (1H,  $dd$ ,  $J = 1.5$  and 4 Hz), 6.72 and 7.20 ( $\text{A}_2\text{B}_2$ ,  $J = 8$  Hz), MS  $m/z$  216 [ $\text{M}$ ] $^+$

Ethyl 3-( $\alpha$ -furoyl)-3-(*p*-methoxyphenyl)-propionate (9) To a suspension of powdered Na (40 mg) in dry toluene (0.5 ml) and dry benzene (3 ml), compound 8 (262 mg) was added and the mixture refluxed for 6 hr Ethyl bromoacetate (200 mg) was slowly added to the stirred mixture and the whole refluxed with stirring for 9 hr After cooling,  $\text{H}_2\text{O}$  was added and the benzene soln separated, dried and evapd CC of the residue on silica gel with  $\text{CHCl}_3$ -hexane (1:1) afforded 9 (71 mg), an oil, IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  1725, 1670, 1605, 1560, 1510, 1250, 880, 830,  $^1\text{H NMR}$  (60 MHz,  $\text{CDCl}_3$ )  $\delta$  1.13 (3H,  $t$ ,  $J = 7$  Hz), 2.78, 3.24 and 4.97 (ABX,  $J_{\text{AB}} = 17$  Hz,  $J_{\text{AX}} = 6$  Hz and  $J_{\text{BX}} = 9$  Hz), 3.85 (3H, s), 4.18 (2H,  $q$ ,  $J = 7$  Hz), 6.52 (1H,  $dd$ ,  $J = 1.5$  and 4 Hz), 6.89 and 7.31 ( $\text{A}_2\text{B}_2$ ,  $J = 8$  Hz), 7.25 (1H,  $m$ ), 7.60 (1H,  $m$ ), MS  $m/z$  302 [ $\text{M}$ ] $^+$

Methyl 3-( $\alpha$ -furoyl)-3-(*p*-methoxyphenyl)-propionate (7) Compound 5 (8 mg) was treated with  $\text{CH}_2\text{N}_2$  to give a residue, which was subjected to CC on silica gel with  $\text{CHCl}_3$ , yielding the Me ether 7 (4.8 mg), an oil,  $[\alpha]_{\text{D}} -35.4^\circ$  (MeOH,  $c$  0.24), IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  1735, 1670, 1605, 1560, 1510, 1250, 880, 830,  $^1\text{H NMR}$  (60 MHz,  $\text{CDCl}_3$ )  $\delta$  2.80, 3.40 and 4.90 (ABX,  $J_{\text{AB}} = 17$  Hz,  $J_{\text{AX}} = 6$  Hz and  $J_{\text{BX}} = 9$  Hz, H-2  $\times$  2 and H-3), 3.72 (3H, s, COOMe), 3.83 (3H, s, OMe), 6.50 (1H,  $dd$ ,  $J = 1.5$  and 4 Hz, furan H-4), 6.86 and 7.27 ( $\text{A}_2\text{B}_2$ ,  $J = 8$  Hz, Ar-H), 7.23 (1H,  $m$ , furan H-3), 7.57 (1H,  $m$ , furan H-5), MS  $m/z$  288 [ $\text{M}$ ] $^+$  A soln of 9 (60 mg) was added to 1 N NaOH (0.5 ml) and the soln stirred at  $85^\circ$  for 1 hr The reaction mixture was diluted with  $\text{H}_2\text{O}$  and extracted with  $\text{Et}_2\text{O}$  to remove neutral material The aq layer was acidified with dil HCl, extracted with  $\text{Et}_2\text{O}$ , washed with  $\text{H}_2\text{O}$  and dried ( $\text{Na}_2\text{SO}_4$ ) The crude product was methylated with  $\text{CH}_2\text{N}_2$  to give a residue CC of the residue on silica gel with  $\text{CHCl}_3$  afforded the Me ester 7 (10 mg), an oil, IR  $\nu_{\text{max}}^{\text{film}}$   $\text{cm}^{-1}$  1735, 1670, 1605, 1565, 1510, 1250, 880, 830,  $^1\text{H NMR}$  (60 MHz,  $\text{CDCl}_3$ )  $\delta$  2.80, 3.40 and 4.90 (ABX,  $J_{\text{AB}} = 17$  Hz,  $J_{\text{AX}} = 6$  Hz and  $J_{\text{BX}} = 9$  Hz), 3.72 and 3.83 (3H each, s), 6.38 (1H,  $dd$ ,  $J = 1.5$  and 4 Hz), 6.91 and 7.32 ( $\text{A}_2\text{B}_2$ ,  $J = 8$  Hz), 7.23 (1H,  $m$ ), 7.60 (1H,  $m$ ), MS  $m/z$  288 [ $\text{M}$ ] $^+$

Methylation of 1 with  $\text{CH}_2\text{N}_2$  followed by oxidation with  $\text{NaIO}_4$  A soln of 1 (160 mg) in MeOH (1 ml) was treated with excess  $\text{CH}_2\text{N}_2$  The crude product (82 mg) was boiled for 3 min in

1 N KOH (1.5 ml) and diluted with  $\text{H}_2\text{O}$  (8.3 ml) The soln was treated with  $\text{NaIO}_4$  (480 mg) in  $\text{H}_2\text{O}$  (8.3 ml) and stirred at room temp for 4 hr The reaction mixture (pH 6.86) was acidified with dil  $\text{H}_2\text{SO}_4$ , continuously extracted with  $\text{Et}_2\text{O}$  and washed with  $\text{Na}_2\text{S}_2\text{O}_3$  and  $\text{H}_2\text{O}$  The crude product was recrystallized from HOAc to yield *p*-methoxyphenylsuccinic acid (2.2 mg), needles, mp  $198$ – $200^\circ$ ,  $[\alpha]_{\text{D}} -15^\circ$  (MeOH,  $c$  0.05), IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$  2740–2600, 1700, 1610, 1580, 1512, 930, 735 The IR spectrum was identical with that of an authentic sample of *p*-methoxyphenylsuccinic acid

Sitosterol Plates from MeOH, mp  $139$ – $140^\circ$ , IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  3300, 1640,  $^1\text{H NMR}$  (60 MHz,  $\text{CDCl}_3$ )  $\delta$  0.68–2.36 ( $m$ ),  $ca$  3.40 (1H,  $m$ ), 5.36 (1H,  $m$ ), MS  $m/z$  414 [ $\text{M}$ ] $^+$  (Found C, 83.78, H, 12.52% Calc for  $\text{C}_{29}\text{H}_{50}\text{O}$  C, 83.99, H, 12.15%) The spectral data were identical with those of an authentic sample

Ursolic acid A white powder from MeOH, mp  $270$ – $288^\circ$ , IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  3400, 1690,  $^1\text{H NMR}$  (60 MHz,  $\text{C}_5\text{D}_5\text{N}$ )  $\delta$  0.93–2.76 ( $m$ ), 3.50 (1H,  $m$ ), 5.67 (1H,  $m$ ) The spectral data were identical with those of an authentic sample

Sitosteryl  $\beta$ -D-glucoside A white powder from MeOH, mp  $263$ – $265^\circ$ , IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  3400, 1650,  $^1\text{H NMR}$  (60 MHz,  $\text{C}_5\text{D}_5\text{N}$ )  $\delta$  0.65–2.63 ( $m$ ), 3.8–4.57 (6H,  $m$ ), 5.08 (1H,  $d$ ,  $J = 8$  Hz), 5.43 (1H,  $m$ ) (Found C, 71.98, H, 10.90% Calc for  $\text{C}_{35}\text{H}_{60}\text{O}_6 \cdot \frac{1}{2} \text{H}_2\text{O}$  C, 72.06, H, 10.54%) The spectral data were identical with those of an authentic sample

Isoquercitrin Needles from MeOH, mp  $209$ – $210^\circ$ , IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  3250, 1650, 1600, 1550, 1505,  $^1\text{H NMR}$  (60 MHz,  $\text{C}_5\text{D}_5\text{N}$ )  $\delta$  3.92–4.42 (6H,  $m$ ), 6.08 (1H,  $d$ ,  $J = 8$  Hz), 6.62 (2H,  $m$ ), 7.22 (1H,  $d$ ,  $J = 8$  Hz), 7.93 (1H,  $dd$ ,  $J = 2$  and 8 Hz), 8.37 (1H,  $d$ ,  $J = 2$  Hz) The spectral data were identical with those of an authentic sample

*p*-Hydroxyphenyl  $\beta$ -D-alloside Prisms from MeOH, mp  $188$ – $191^\circ$  and mp  $201$ – $202^\circ$ , IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  3200, 1600, 1510, 1210, 1090, 1040, 870, 850, 830, 790,  $^1\text{H NMR}$  (60 MHz,  $\text{C}_5\text{D}_5\text{N}$ )  $\delta$  4.10–4.73 (5H,  $m$ ), 4.83 (1H,  $t$ -like,  $J = 3$  Hz), 5.97 (1H,  $d$ ,  $J = 8$  Hz), 7.06 and 7.32 ( $\text{A}_2\text{B}_2$ ,  $J = 9$  Hz) The spectral data were identical with those of an authentic sample

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